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Key Points:

- The main factors that influenced wetland greenhouse gas emissions were plant growth, water flow, and nitrogen loading
- The restored wetland was a much stronger carbon sink than the natural wetland by maximizing CO₂ uptake and minimizing CH₄ emissions
- Altering species composition and water flow as part of wetland restoration was an innovative strategy for mitigating climate change

Supporting Information:

- Supporting Information S1

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Enhanced Carbon Uptake and Reduced Methane Emissions in a Newly Restored Wetland

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Abstract Wetlands play an important role in reducing global warming potential in response to global climate change. Unfortunately, due to the effects of human disturbance and natural erosion, wetlands are facing global extinction. It is essential to implement engineering measures to restore damaged wetlands. However, the carbon sink capacity of restored wetlands is unclear. We examined the seasonal change of greenhouse gas emissions in both restored wetland and natural wetland and then evaluated the carbon sequestration capacity of the restored wetland. We found that (1) the carbon sink capacity of the restored wetland showed clear daily and seasonal change, which was affected by light intensity, air temperature, and vegetation growth, and (2) the annual daytime (8–18 hr) sustained-flux global warming potential was $-11.23 \pm 4.34 \text{ kg CO}_2 \text{ m}^{-2} \text{ y}^{-1}$, representing a much larger carbon sink than natural wetland ($-5.04 \pm 3.73 \text{ kg CO}_2 \text{ m}^{-2} \text{ y}^{-1}$) from April to December. In addition, the results showed that appropriate tidal flow management may help to reduce CH₄ emission in wetland restoration. Thus, we proposed that the restored coastal wetland, via effective engineering measures, reliably acted as a large net carbon sink and has the potential to help mitigate climate change.

1. Introduction

Currently, substantial greenhouse gas (GHG) emissions, including carbon dioxide (CO₂), methane (CH₄), and nitrous oxide (N₂O), have contributed to global climate change, which is a threat to human survival and development (Intergovernmental Panel on Climate Change, 2014). It is urgently needed to explore more approaches to fix carbon and reduce GHG emissions to mitigate climate change progression (National Academy of Sciences, 2018).

Coastal wetlands (e.g., salt marshes), the ecological buffer zones between the land and ocean, are large soil carbon stocks that can maintain rapid carbon sequestration (therefore significant blue carbon ecosystems) with high photosynthetic capacity and low decomposition rates (Tang et al., 2018). Compared to terrestrial carbon sinks (e.g., forests), blue carbon ecosystems sequester 10 times or more carbon than terrestrial ecosystems due to the long-term storage of carbon in sediments (McLeod et al., 2011). Therefore, protecting and maintaining vulnerable but important wetland carbon stocks to avoid emissions and the release of stored carbon are important approaches for climate change mitigation efforts.

Natural wetlands have shown tolerance when confronted with human impacts and climate change effects (Kirwan et al., 2016; Kirwan & Megonigal, 2013). However, in some severely damaged wetlands, vulnerability to environmental stressors is high and the effects may be irreversible. These large-scale coastal wetlands and their associated ecosystem services have declined due to the combined effects of human disturbance and natural erosion (Mitsch & Gosselink, 2000).

Previously, coastal wetland restoration implementations mainly focused on restoring tidal marshes by introducing tidal flow, amending sediment, and transplanting plants (Mitsch & Gosselink, 2000). Recently, as blue carbon has attracted global attention (supporting information, Table S1), coastal wetland restoration projects are expected to incorporate methods to enhance carbon sequestration that go beyond just restoring marshes (Mander et al., 2015; Zhou et al., 2016). Unfortunately, up until now, most researchers focused on the water quality and biodiversity and rarely quantified blue carbon potential of restored wetlands. Nevertheless, most blue carbon studies on coastal wetlands mainly focused on CO₂ sequestration and

often ignored CH₄ and N₂O emissions (Kroeger et al., 2017). CH₄ and N₂O have larger global warming potentials per molecule, about 25 and 298 times higher than that of CO₂, respectively (Brannon et al., 2016). Thus, if the restored wetland increases CH₄ or N₂O emissions, it alters the overall benefit of CO₂ sequestration (García-Lledó et al., 2011; Rosentreter et al., 2018).

In this study, in situ GHG measurements (CO₂, CH₄, and N₂O) were simultaneously collected in the field to accurately estimate the blue carbon storage of the whole ecosystem in the coastal restored and natural wetlands, which may be offset by CH₄ and N₂O emissions. The aim is to discuss the global warming potential and accurately assess the ecosystem carbon service (CO₂ uptake capacity and CH₄ emissions) of the restored wetland via effective engineering measures. The results will lead to a greater understanding of blue carbon ecology and promote the implementation of wetland restoration worldwide.

2. Materials and Methods

2.1. Study Sites

This study was conducted within the coastal wetlands of northern Hangzhou Bay, Shanghai, China. The Yingwuzhou wetland (restored wetland) and the Fengxian wetland (natural wetland) were used to explore the blue carbon effect of the restored wetland throughout the year of 2018. These two study sites were 20 km apart (Figure S1). The Yingwuzhou wetland and the Fengxian wetland cover an area of 23 and 8,000 ha, respectively, and are representative of the restored and natural estuarine wetlands in southeast China.

Yingwuzhou wetland, located in Jinshan District in Shanghai, China (N30°42′26.73″, E121°20′04.15″), was restored in 2016 by amending the sediment, constructing an ecological dam to prevent erosion, and transplanting local plants. This restored wetland provided ecological benefits of water quality improvement and biodiversity protection. Before restoration, the Yingwuzhou wetland was a *Phragmites australis* and *Spartina alterniflora* dominated wetland, like the Fengxian natural wetland. To protect the native species—*Phragmites australis*, as one of the major goals of the wetland restoration in Shanghai, we just transplanted *Phragmites australis* but not *Spartina alterniflora* in Yingwuzhou wetland. The salinity and pH of the wetland water were $10 \pm 1.22\text{‰}$ and 6.40 ± 0.13 , respectively. A culvert was designed to connect the inside wetland and outside tidal water. Consequently, though there is no tidal flooding, we could control water exchange and maintain a desired water level by controlling the valve of the culvert (Figure S2).

The natural wetland site is located in Fengxian District, Shanghai (N30°50′33.75″, E121°39′03.60″). The dominant species are *Phragmites australis* and *Spartina alterniflora* in this area, which were flooded by high tide on the first and 15th of every lunar month. The salinity and pH of the wetland water were $19.28 \pm 2.12\text{‰}$ and 6.45 ± 0.12 , respectively.

2.2. GHG Flux Measurements

At the restored and natural wetland site, five plots (2 × 2 m, the plots were 8 m apart from each other) were installed at each site during the winter of 2017. Within each plot, three PVC rings (20-cm height × 20-cm diameter, and installed into the soil to a depth of 10 cm) were placed, which represented the three replicates per plot that were established in both the restored and natural wetlands.

CO₂ and CH₄ fluxes were measured simultaneously in situ throughout the year of 2018, using an Ultra-Portable GHG Analyzer (Los Gatos Inc., CA, USA). A 2-m-high, 0.2-m-diameter transparent polycarbonate chamber was used to accommodate the height of the plants (up to 2 m) inside the chamber, with a mini-fan inside the chamber to ensure air circulation during measurements. Nylon tubing (0.46-cm inner diameter and approximately 10 m in total length) was connected to the analyzer via two gastight ports in a closed loop. Gas measurements were conducted for 5–10 min per plot, based on observed periods for linear rates of change. Before the GHG measurements, we made modifications to the instrument and kept the static transparent chamber be gastight in order to accomplish a closed-loop static flux chamber.

The N₂O concentration measurement was performed by adopting a closed static chamber-gas chromatograph technique (Brannon et al., 2016). The gas samples were drawn from the static transparent chamber into 60-ml nylon syringe at 0, 5, 10, 15, and 20 min and then transferred to the 50-ml vacuum airbag (MBT41-0.1, Haide Technologies Co. Ltd., Dalian, China; Brannon et al., 2016). The concentration of N₂O was measured by the gas chromatograph (7890A, Agilent Technologies Co. Ltd., CA, USA) with the

electronic capture detector. The column temperature and the detector temperature were set as 60 and 330 °C, respectively, and the carrier gas flow rate was 10 ml min⁻¹.

To draw monthly comparisons of the restored and natural wetlands at similar environmental conditions (e.g., light intensity, temperature, and plant phenology), all GHG flux measurements were performed between 10:00 and 15:00 on clear, sunny days to quantify the random error (i.e., daily difference; Lavoie et al., 2014). There was a minimal monitoring interval between the two sites (usually, measurements were finished within 2 days, unless there was inclement weather).

2.3. GHG Fluxes and Sustained-Flux Global Warming Potential Calculations

GHG fluxes were calculated using the linear change in gas concentrations over time with field-measured air temperatures and atmospheric pressure (Martin & Moseman-Valtierra, 2017) during the measurement period, following the formula:

$$F = \frac{dC}{dt} \times \frac{PV}{RAT}, \quad (1)$$

where F is the GHG flux ($\mu\text{mol m}^{-2} \text{s}^{-1}$); dC/dt is the changing concentration over time ($\mu\text{mol mol}^{-1}$); P is the air pressure, standard is 101,223.7 (Pa); V is the effective volume of the static closed chamber (m^3); R is the gas constant, defaulted to 8.3144 (J mol K); A is the base area of the chamber (m^2); and T is the air temperature (K). For CO_2/CH_4 data analysis, the first 30 s and the end 30 s of measurements were not included in the flux calculations to account for gases passing through the length of the tubing between the analyzer and the chamber. When dC/dt had an R^2 value of less than 0.9 and p value greater than 0.05, data were not included in the analysis.

Radiative forcing (W m^{-2}) is a useful measurement to estimate the potential climatic effect induced by the changing concentrations of radiatively active (greenhouse) gases, solar radiation, aerosols, and albedo (Boucher & Haywood, 2001; Froking et al., 2006; Jain et al., 2000). In this study, we used sustained-flux global warming potential (SGWP) to estimate the dynamics of total radiative forcing as ecosystem GHG fluxes are sustained ones. SGWP was calculated with scalers of 45 for CH_4 and 270 for N_2O considering the warming effect over the 100-year period (Neubauer & Megonigal, 2015). Because of lack of nighttime flux measurement, here we only calculate daytime (8–18 hr) SGWP scaled to a day or a year. In addition, we estimated the annual average SGWP from April to December in the restored and natural wetlands.

$$\text{SGWP} = F\text{CO}_2 + (F\text{CH}_4 \times 45) + (F\text{N}_2\text{O} \times 270). \quad (2)$$

where $F\text{CO}_2$, $F\text{CH}_4$, $F\text{N}_2\text{O}$ are mass flux in units (for example, $\mu\text{g CO}_2 \text{m}^{-2} \text{s}^{-1}$).

2.4. Plant Physiological Properties

To obtain the plants' chlorophyll concentration measurement, we used a Soil Plant Analysis Development-502 meter. This equipment is widely used as a nondestructive, inexpensive, rapid, and accurate tool that utilizes leaf transmittance in two wavebands centered at 650 and 940 nm (Yang et al., 2017). We selected five leaves per plot (25 leaves per site). For each leaf, we made five Soil Plant Analysis Development readings, which were evenly distributed over the entire leaf area and then averaged.

In the summer, we investigated the plant biomass (the live vegetation biomass per unit area) in the restored and natural wetlands. In each site, we randomly set seven quadrats ($1 \times 1 \text{ m}$) near the plots and cut the vegetation (aboveground biomass) in the quadrats. Then the vegetation was taken back to the lab and dried in the oven (65 °C) for at least 48 hr to quantify leaf dry mass.

2.5. Key Environmental Parameters

The environmental data were obtained from the Micro Weather Station (Onset-U30, USA). The data logger collected the following measurements every 5 min: photosynthetically active radiation ($\mu\text{mol m}^{-2} \text{s}^{-1}$), air temperature (T_{air} , °C), relative humidity (%), and so forth. Soil pH, oxidation-reduction potential (ORP), temperature, moisture, and water salinity point measurements were performed monthly in each PVC ring. Soil ORP and pH were measured using an ORP meter (FJA-6, Nanjing, China), soil temperature and moisture were measured by Decagon EC-5 sensors (Decagon Devices, Pullman, WA, USA), and water salinity was

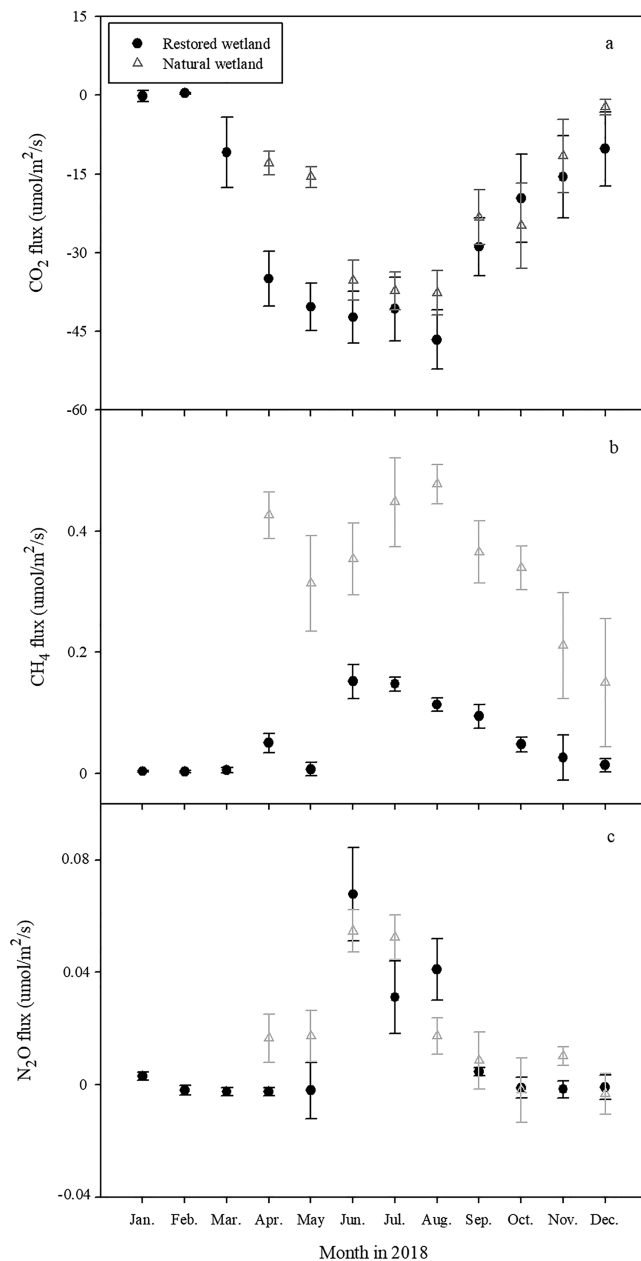


Figure 1. The seasonal variations of GHG emissions (mean \pm SD) in the restored and natural wetlands.

measured by a handheld refractometer (MASTER-S/Mill α , ATAGO, Japan) by applying water extracted from approximately 5 ml of surface soil.

The soil samples used to obtain soil organic carbon (SOC) were collected from the cores centered at 15-cm depth in August. Three soil subsamples were randomly collected and then mixed as one sample in each PVC ring. After collection, we separated undecomposed dead plant matter from the samples and placed the samples in a cooler and taken back to the lab. After they were dried and ground into powder, the soil samples were filtered through a 150- μ m mesh screen. The soil samples were acidified with 1 mol L⁻¹ of HCl to remove the inorganic carbon (Vuong et al., 2013). Then, the samples were ground again, and 2 mg of the sample was used for SOC determination by an Elemental Analyzer (Vario MICRO cube, Elementar, Germany).

For the dissolved inorganic nitrogen (DIN) analysis, the Rhizon soil moisture samplers (Rhizosphere Research Products, Wageningen, Netherlands) were used to sample the pore water (50 ml in each plot) within a 15- to 20-cm soil depth (quarterly in 2018). The water samples were immediately placed in a cooler and transported back to the lab where the samples were filtered and then tested for key nitrogen indicators in the ultraviolet and visible spectrophotometer (UV-7504, Xinmao Co. Ltd., Shanghai, China). The concentrations of NO₃⁻-N, NO₂⁻-N, and NH₄⁺-N were analyzed by the zinc cadmium reduction method, naphthylethylenediamine photometric method, and hypobromite oxidation method, respectively; thus, DIN concentration was calculated as the sum of the values of NO₃⁻-N, NO₂⁻-N, and NH₄⁺-N.

2.6. Statistical Analysis

We conducted statistical analyses using SPSS 11.0 (SPSS Inc., Chicago, IL, USA), and a significance level of $p < 0.05$ was used throughout. The scaling error (spatial variability) was defined as standard deviation of the GHG emissions measurements made on the same day (Lavoie et al., 2014; Richardson & Hollinger, 2005). The significant differences of GHG emissions and SGWP between the sites through time were assessed by a repeated measures ANOVA test. The significance between sample size and GHG fluxes variation in each month of the different wetland sites was analyzed by test of coefficient of variation. We used regression analyses to evaluate the influence of leaf chlorophyll concentration and light intensity on CO₂ emission fluxes, and the influence of soil ORP on CH₄ emission fluxes, which was judged statistically by the coefficient of determination, R^2 , and its statistical significance was determined by one-way ANOVA.

3. Results

3.1. CO₂ Fluxes in the Restored and Natural Wetlands

The CO₂ fluxes represented daytime gross primary production with the negative value indicating CO₂ uptake and the positive value CO₂ release (Figure 1a). Overall, both the restored and natural wetlands showed clear seasonal patterns of CO₂ uptake during the growing season (Figure 1a).

In the restored wetland, while the plants were still dormant, the CO₂ uptake ability remained low. Especially in February, the average CO₂ flux values were positive (0.312 μ mol m⁻² s⁻¹), indicating a small CO₂ release from the wetland (Figure 1a). In March, plants displayed daytime CO₂ uptake (the CO₂ flux values were negative), then the uptake increased in spring, and maintained high values (40–47 μ mol m⁻² s⁻¹) in the

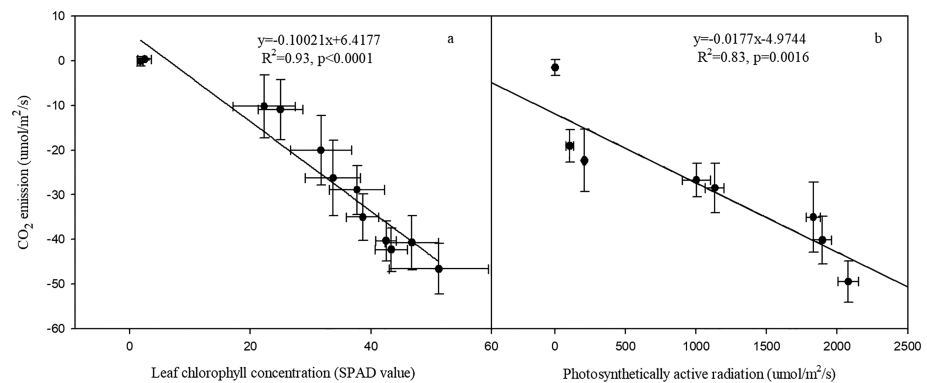


Figure 2. The linear relationships (a) between CO₂ emission fluxes and leaf chlorophyll concentration and (b) between CO₂ emission fluxes and the light intensity of the restored wetland.

summer coinciding with plant growth, which indicates the strong gross primary production of the wetland ecosystem. In the autumn, the plant photosynthetic capacity began to decrease and reached to a low level with plant senescing. There was a significant linear correlation between leaf chlorophyll concentration and CO₂ uptake ($R^2 = 0.93$, $p < 0.0001$; Figure 2a), which could explain the seasonal change of CO₂ uptake with the shifts of plant phenology. Since the correlation between CO₂ emission fluxes and leaf chlorophyll concentration in the natural wetland is similar to that in the restored wetland, we just plotted the correlation of the restored wetland. The CO₂ fluxes of natural wetland also showed seasonal change with the development of plants. However, the CO₂ uptake ability of natural wetland over the year (April–December) was significantly lower than that of restored wetland (Figure 1a; $p < 0.0001$, Table S2).

In addition, plant CO₂ uptake ability appeared to be influenced by daily changes in weather (sunny vs. cloudy). Diurnal measurements made on a midsummer day showed that plant CO₂ uptake was correlated with photosynthetically active radiation ($R^2 = 0.83$, $p = 0.0016$; Figures 2b and 3a). In the early morning and later afternoon, the plants showed low photosynthetic rate under low light intensity; plant CO₂ uptake ability became stronger with the increasing of light intensity and peaked (uptake $48 \text{ CO}_2 \mu\text{mol m}^{-2} \text{ s}^{-1}$) at 14:00 with the greatest lighting. At night, plants ceased photosynthesis, and whole ecosystem dark respiration at 20:00 averaged $6.81 \text{ CO}_2 \mu\text{mol m}^{-2} \text{ s}^{-1}$. We found that the CO₂ fluxes were relatively stable during 10:00–15:00, so subsequent campaigns were conducted during this midday time window.

3.2. CH₄ Emission in the Restored and Natural Wetlands

Generally, both the restored and natural wetlands represented CH₄ sources throughout the year. Compared with CO₂ flux, the CH₄ emission level of the wetlands was two order of magnitude lower.

In the restored wetland, before the active growing season began, CH₄ fluxes were very low (ranged from 0 to $0.05 \mu\text{mol m}^{-2} \text{ s}^{-1}$). In April, with rising temperature and plant growth, CH₄ flux increased; the peak of CH₄ emission reached $0.15 \mu\text{mol m}^{-2} \text{ s}^{-1}$ in June. The CH₄ emission remained high in the summer and then reduced after entering the senescence stage (Figure 1b). There was no clear change of CH₄ emission during the day, which indicates that light intensity may not be the key environmental factor controlling CH₄ emission in the restored wetland (Figure 3b).

Comparatively, CH₄ emission fluxes of natural wetland were significantly higher than that of restored wetland ($p < 0.0001$, Table S2) and varied seasonally (increased with plants growth and decreased with plants aging). The CH₄ emission of the natural wetland peaked in August with the value of $0.48 \mu\text{mol m}^{-2} \text{ s}^{-1}$, which was four times higher than the maximum of restored wetland.

3.3. N₂O Emission in the Restored and Natural Wetlands

Similar to CH₄ fluxes, the N₂O emission values were positive, which revealed that the wetlands were N₂O sources. Figure 1c showed the seasonal patterns of N₂O fluxes in the restored and natural wetlands. In spring and winter, the uptake of N₂O within the restored wetland was minimal. Then, the N₂O fluxes increased substantially and peaked in June (N₂O fluxes were 0.07 and $0.06 \mu\text{mol m}^{-2} \text{ s}^{-1}$ in the restored and

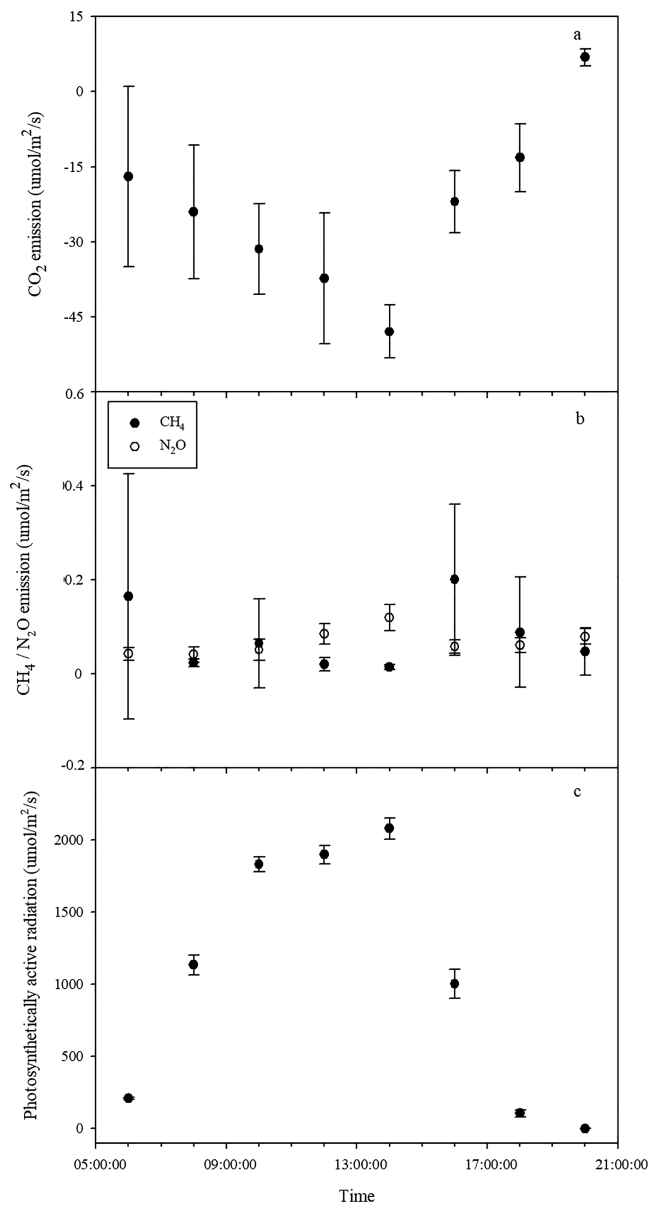


Figure 3. The diurnal variations of GHG emissions (mean \pm SD) of the restored wetland on 15 July 2018.

natural wetlands, respectively). In late summer and autumn, the values dropped drastically, close to 0 in October. Similar to CH_4 , N_2O flux had no notable correlation to photosynthetically active radiation in summer, with a range of $0.04\text{--}0.12 \mu\text{mol m}^{-2} \text{s}^{-1}$ (Figure 3c). There was no significant difference of the N_2O emission fluxes between the natural and restored wetlands ($p = 0.073$, Table S2).

3.4. The SGWP in the Restored and Natural Wetlands

Generally, the SGWP values of the restored and natural wetlands were negative, which indicated the wetlands could reduce the GHG emissions and potentially help to mitigate climate change progression. The lowest SGWP value appeared in May with the value of $-0.065 \text{ kg CO}_2 \text{ m}^{-2} \text{ day}^{-1}$, which showed that the restored wetland had the strongest blue carbon potential at that time. In June, the SGWP of the restored wetland increased suddenly potentially due to the contribution of N_2O emission (Figures 1c and 4). Although the flux of N_2O was orders of magnitude lower than that of CO_2 and CH_4 (Figure 1), small changes of N_2O emission can shift the SGWP value of the wetland ecosystem.

Most notably, this study showed that the restored wetland had significantly lower SGWP compared to the natural wetland (Figure 4; $p < 0.0001$, Table S2). The annual daytime SGWP value of the restored wetland was approximately 2.2 times that of the natural wetland (SGWP values were -11.23 and $-5.04 \text{ kg CO}_2 \text{ m}^{-2} \text{ year}^{-1}$ from April to December in the restored and natural wetlands, respectively, Table 1), which indicates the restored wetland had a higher capability to mitigate climate change.

4. Discussion

4.1. GHG Fluxes in Restored and Natural Wetlands

In this study, greater CO_2 absorption of the restored wetlands relative to the natural wetlands was due to greater photosynthetic uptake by greater plants biomass (Windham et al., 2001). In the restored wetland, the plants grew well after transplantation, and the biomass was significantly higher ($p = 0.042$), leading to more CO_2 uptake in the restored wetland than that in the natural wetland, which reflected the success of the restoration plan (Table 1). This result is potentially due to the creation of a suitable micro-environment that facilitates rapid plant growth via “culvert valve control” in the restored wetland. After marsh establishment was completed, the eco-dike was opened to enable tidal access, thus further encouraging a self-sustaining wetland to promote ecosystem functions and services.

Meanwhile, we found that CO_2 uptake of restored wetland was significantly higher than natural wetland throughout the year, except in November (Figure 1a). During the dormant period (October and November), the spatial variation (i.e., standard deviation) across the plots were higher in both sites (Figure 1), indicating that there were “hotspots” and “hot moments” caused by unclear environmental factors in the non-growing season (He et al., 2010; Lavoie et al., 2014). Though our monthly measurements of the plots could represent the seasonal change of GHG emissions in the two sites (Table S3), the large flux uncertainty may affect evaluating the monthly integrated fluxes. Further work is needed to reduce the uncertainty in temporally integrated flux by increasing measurement frequency (He et al., 2010) or applying gap-filling algorithms when scaling estimates are derived from sporadic manual measurements in time as well as space (Gomez-Casanovas et al., 2013; Phillips et al., 2017; Richardson & Hollinger, 2005).

Additionally, the restored wetland emitted less CH_4 compared with the natural wetland, which may due to the ORP caused by different water managements (Table 1 and Figures 5 and 6). Mostly, the natural wetland

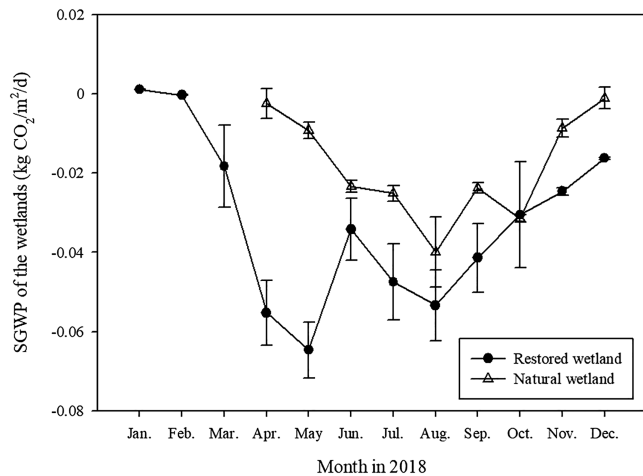


Figure 4. The seasonal variations of SGWP (mean \pm SD) in the restored and natural wetlands. Here, the SGWP value represented the daytime carbon absorption capacity of the wetlands from 8 to 18 hr of a day.

did not undergo water exchange from low tide and consequently formed an anoxic condition and generated CH₄ (Figure 5). Therefore, in the restored wetland, we kept the water level at 10 cm and the flow rate at 0.03 m s⁻¹ throughout the year. The manipulation of tidal flow can effectively reduce the CH₄ emission due to continuous water flow that disrupts the anoxic environment, thus inhibiting methanogenesis (Altor & Mitsch, 2006). Consequently, coastal ecosystems could switch from net sources to net sinks of carbon with efficient water exchange. Therefore, including tidal flow management in future wetland restoration/creation may be a feasible method for reducing CH₄ emission.

To determine the effect of water management on CH₄ emissions, we shut down the flow pump and evacuated the overlying water within the restored wetland for 2 weeks (13 August–28 August) to simulate the natural wetland under low tide. We found that continuous flow enhanced the ORP of water-soil interface, consequently decreasing CH₄ emission in the restored wetland (Figure 5).

Previous studies also stated that CH₄ emission was negatively related to wetland salinity (Chmura et al., 2011; Kroeger et al., 2017; Poffenbarger et al., 2011; Vivanco et al., 2015) because (1) sulfate-reducing bacteria outcompete methanogens for energy sources and then limiting CH₄ production (Poffenbarger et al., 2011; Weston et al., 2014) and (2) CH₄ oxidation by sulfate reducers can also inhibit CH₄ emission (Bartlett et al., 1987). However, our results showed that the natural wetland emitted higher CH₄, though it had a higher salinity level than the restored wetland (Table 1), which suggests that salinity is not the dominant factor influencing CH₄ emission in all moderate saline wetlands.

4.2. The Key Factors Affecting the Seasonal Dynamics of GHG Fluxes in Wetlands

Plant growth is important for carbon sequestration and preservation due to atmospheric CO₂ uptake from coastal wetland plants (Couto et al., 2014). There was a clear linkage between plant growth (leaf chlorophyll concentration) and daytime CO₂ uptake during the growing season, suggesting a strong role of vegetation photosynthesis in reducing GHG emission and enhancing C sequestration. Though *Phragmites australis* and *Spartina alterniflora* may have no significant difference in photosynthetic capacity with the similar leaf

Table 1

The Structural and Functional Properties of Blue Carbon in the Restored and Natural Wetlands

GHG fluxes and environmental variables	Restored wetland	Natural wetland
CO ₂ emission ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	-43.22 ± 3.03	-36.75 ± 1.30
CH ₄ emission ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	0.13 ± 0.02	0.43 ± 0.07
N ₂ O emission ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	0.05 ± 0.02	0.04 ± 0.02
SGWP ($\text{kg CO}_2 \text{m}^{-2} \text{year}^{-1}$)	-11.23 ± 4.34	-5.04 ± 3.73
pH	6.40 ± 0.13	6.45 ± 0.12
ORP (mV)	301.37 ± 52.77	203.33 ± 32.61
Soil moisture (%)	90.15 ± 5.84	82.93 ± 6.53
Water salinity (‰)	10.00 ± 1.22	19.28 ± 2.12
SOC content (%)	1.01 ± 0.08	0.93 ± 0.13
DIN concentration (mg L^{-1})	0.16 ± 0.01	0.18 ± 0.01
Plant composition	<i>Phragmites australis</i>	<i>Phragmites australis</i> (45%) and <i>Spartina alterniflora</i> (55%)
Plant biomass (kg m^{-2})	2.25 ± 0.19	2.04 ± 0.12
Water management	The water level was 10 cm, and the continuous flow rate was 0.03 m s ⁻¹	Natural tide effect

Note. The SGWP value represented the annual sum (mean \pm SD) calculated as daytime SGWP from 8 to 18 hr over 275 days (April–December), and other data represented the averaged values (mean \pm SD) of the measurements during plants' maturation stage in the summer season (June–August).

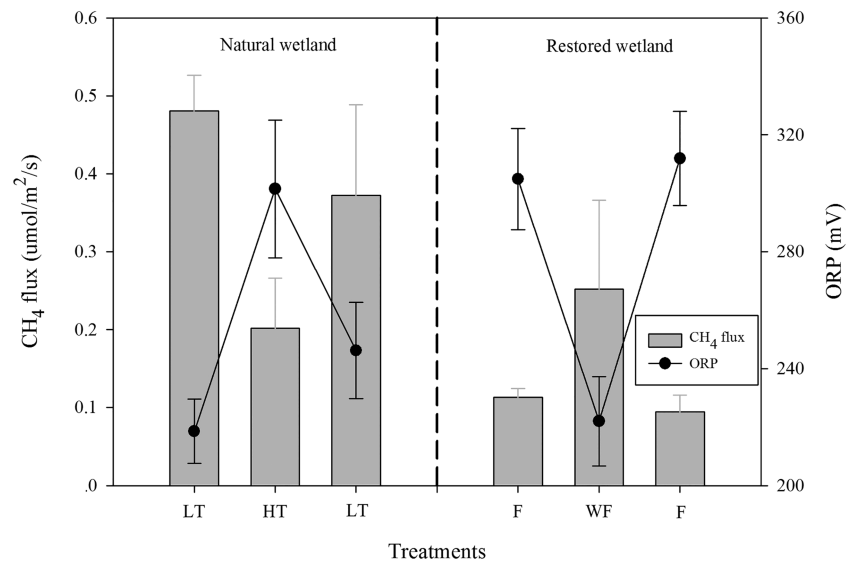


Figure 5. The CH₄ emissions and ORP of the restored and natural wetlands. In the natural wetland, measurements were taken on 3 July (low tide, LT), 13 July (high tide, HT), and 20 July (LT); in the restored wetland, measurements were taken on 10 August (flow rate of 0.03 m s⁻¹, F), 28 August (without flow, WF), and 5 September (flow rate of 0.03 m s⁻¹, F).

chlorophyll concentrations, Yingwuzhou wetland showed stronger carbon assimilation along with higher plant biomass than the natural wetland. However, despite the *Phragmites australis* biomass has a steady increase in its early establishment phase, the plant growth rate might decrease over years as the ecosystem further established in the future. CH₄ emission increased in the summer months, which is a similar pattern to plant growth as it increased during the same time period (Colmer, 2003; Koebsch et al., 2013). There are three main reasons that account for this phenomenon: (1) root respiration could accelerate O₂ consumption and anaerobic environment formation; (2) root exudates provide sufficient substrate for methanogens (Chidthaisong & Watanabe, 1997); and (3) plants' aerenchyma can also directly drive CH₄ transferring from soils to the atmosphere (Beckett et al., 2001; Colmer, 2003).

Temperature and light intensity are the important environmental drivers of GHG fluxes in wetlands. The seasonal change of daytime CO₂ uptake by wetland plants is mainly driven by temperature, photosynthetically active radiation, and day-length (Moseman-Valtierra et al., 2016). Previous studies reported that higher

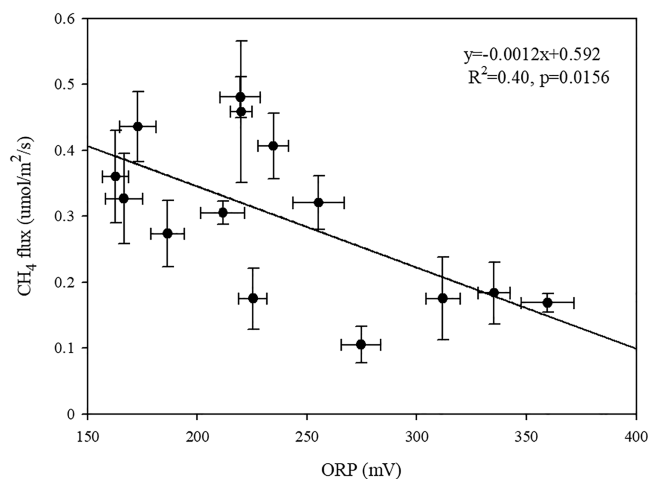


Figure 6. The linear regression demonstrating the influence of ORP on CH₄ emissions of the coastal wetlands in the summer season.

ambient temperature and light intensity could accelerate the activity of the primary C₄ photosynthetic enzyme (RuBisCO), resulting in higher CO₂ uptake (Abdul-Aziz et al., 2018; Guo et al., 2009; Inglett et al., 2012). Moreover, methanogenesis is substantially driven by temperature (Martin & Moseman-Valtierra, 2017), which could contribute to higher CH₄ emission during high temperatures. Walter and Heimann (2000) found that CH₄ production exponentially increased with elevated soil temperature, leading to higher CH₄ emissions. Similarly, the temperature variation may cause the seasonal change of N₂O fluxes. Higher temperatures would benefit the population growth and activity of nitrifiers and denitrifiers (Braker et al., 2010; Gamble et al., 1977), and consequently accelerate nitrification-denitrification processes and generate N₂O from wetland soils (Virdis et al., 2010).

Our results showed that the restored wetland had relatively high N₂O emissions in the summer with high nitrogen loading. The DIN concentration of the pore water was 0.16 ± 0.01 mg L⁻¹ in summer, which was twice higher than other seasons in the restored wetland (the DIN concentrations were 0.08 ± 0.01 , 0.08 ± 0.02 , and 0.09 ± 0.01 mg L⁻¹ in spring,

autumn, and winter, respectively). Previous studies (Gao et al., 2017; Lyu et al., 2017; Moseman-Valtierra et al., 2011) revealed that high nitrogen loading may lead to a considerable amount of N_2O emission from wetlands due to nitrification-denitrification processes, which is consistent with our result. Moreover, given the same DIN level in the pore water (Table 1) of the restored and natural wetlands, we suggested that the DIN level may not be the main factor for the fast plant growth in our study.

4.3. The Evaluation and Comparison of Blue Carbon Capacity

The goal of our study was to analyze GHG emissions and blue carbon capacity of restored coastal wetlands and to support the need for blue carbon capacity to be an important restoration aim and evaluation index for wetland restoration on a global scale. As the results of this study demonstrate, a wetland's blue carbon capacity is an important factor when considering climate change progression, which has implications on human livelihood and development.

In general, the Yingwuzhou restored wetland emerged as a large carbon sink in 2018, and its annual daytime (8–18 hr) SGWP was $-11.70 \pm 7.77 \text{ kg CO}_2 \text{ m}^{-2} \text{ year}^{-1}$ (January–December, or $-11.23 \pm 4.34 \text{ kg CO}_2 \text{ m}^{-2} \text{ year}^{-1}$ if only accounting April to December; Table 1), which would be equivalent to $31.92 \pm 21.20 \text{ MgC ha}^{-1} \text{ year}^{-1}$ for daytime carbon sequestration. This rate exceeded the carbon sequestration rate of eutrophic impoundments at $21.2 \text{ MgC ha}^{-1} \text{ year}^{-1}$ (Downing et al., 2008). Chmura et al. (2003) investigated 154 saline tidal wetlands and obtained a mean carbon sequestration rate of $1.4 \text{ MgC ha}^{-1} \text{ year}^{-1}$, which is also considerably lower than the Yingwuzhou restored wetland because we only consider the daytime carbon uptake capacity in this study without accounting for nighttime respiratory loss of carbon. The ability of reducing GHG emissions was demonstrated more in the restored wetland than the natural wetland, which suggests that restoration of biological and physical conditions (e.g., species composition and water flow) could enhance carbon accumulation rates. These biological and physical conditions can be applied to future wetland restoration projects that occur in various geographical locations as a method to mitigate GHG emissions.

The principle restoration idea is to protect the site and help marsh establishment by engineering measures during the early phases of restoration, and thereafter, in the late phase, we will encourage the self-organization of the restored ecosystem. By proving the strong blue carbon capacity of the wetland restoration, we call for more worldwide implementations of this restoration engineering in degraded wetland areas. Badiou et al. (2011) compared the carbon sequestration potentials in newly (<5 years) restored and long-term (>5 years) restored wetlands in the Canadian prairies and found that the latter was 2.5 times than the former (with the unit as $\text{MgC ha}^{-1} \text{ year}^{-1}$). This study only captured the second year following restoration and compared only a single pair of wetlands in a short-term study. Long-time monitoring of the inter-annual variability of GHG fluxes in more wetland sites is essential to predict GHG fluxes at longer time-scales, which will improve the understanding of the feasibility and effectiveness of wetland restoration to enhance carbon sequestration. The results suggested that the restoration engineering of water flow rates was critical to carbon sequestration, which was substantiated by measurements of CH_4 emissions at different water levels. Hence, we plan to maintain the water level when the restoration process proceeds in the future. However, the long-term impacts of water level on SOC accumulation could not be determined from this short-duration study. It should be noted that GHG emissions from the wetlands themselves do not reflect the total amount of GHGs the wetland contributes to the ecosystem (Poffenbarger et al., 2011), since some fraction of the GHG produced may be transported in water outflow (e.g., DOC and DIC), which future studies should investigate.

5. Conclusion

This study broadened the knowledge of diurnal and seasonal variations of GHG (including CO_2 , CH_4 , and N_2O) emissions, as well as the key environmental factors impacting the vegetated restored wetland. Our experiment revealed that water regulation and adequate plant biomass is emerging as an effective method to promote carbon storage in restored coastal habitats. Our study also quantified a new and previously undervalued strategy for mitigating climate change that maximizes CO_2 uptake and minimizes CH_4 emissions in restored wetlands. The methods that have shown effective in this study can be applied to wetland

restoration projects in other geographical locations as a mechanism to maintain salt marshes as carbon sinks and help mitigate GHG emissions.

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Erratum

In the originally published version of this article, Equation 2 ($SGWP = FCO_2 + 45FCH_4 + 270FN_2O$), FCO_2 , FCH_4 , FN_2O erroneously published as molar flux in units (e.g., $\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$). As a result, Table 1, Figure 4, and Table S2 published incorrectly. Although calculation results were incorrect using this equation, the conclusion of the paper was not influenced.

Corrections were made on the following pages:

Page 1. Abstract: “ -10.32 ± 4.29 ” changed to “ -11.23 ± 4.34 ”

Page 1. Abstract: “ -0.76 ± 3.07 ” changed to “ -5.04 ± 3.73 ”

Page 3. Under Equation (2): “where FCO_2 , FCH_4 , FN_2O are mass flux in units (for example, $\mu\text{gCO}_2 \text{ m}^{-2} \text{ s}^{-1}$)” added

Page 6. “ -0.064 ” changed to “ -0.065 ”

Page 6. “13” to “2.2” “ -10.32 and “ -0.76 ” changed to “ -11.23 and “ -5.04 ”

The article text, Figure 4, Table 1, and Table S2 have since been corrected, and this version may be considered the authoritative version of record.